

and forms at higher temperatures and has a body center crystal structure cubic (bcc). Phase changes α to γ U phases are carried out by heating or adding metal elements at high temperatures then cooled to room temperature. Through the addition of metal elements at high temperatures, the γ U phase formed at high temperatures will remain even though the alloy temperature is cool. A change also follows this phase change in the crystal structure from the original phase α has an orthorhombic structure that will change to the γ U phase, which has a cubic structure. Meanwhile, UO_2 compounds are formed due to the oxidation of the U-Zr-Nb sample either during the hydriding-dehydriding process or after the hydriding-dehydriding process, where the U-Zr-Nb sample is stored in an open place and in contact with air.

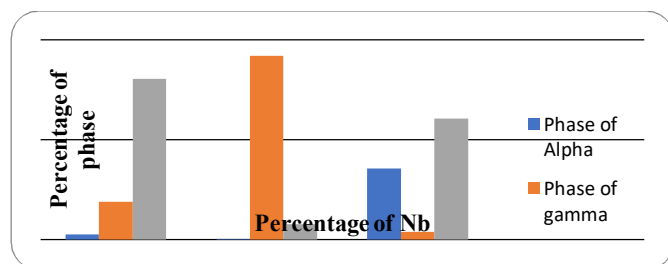


Fig. 6 Phase composition with Nb variation content in the U-Zr-Nb powder

When compared with U-6Zr-2Nb, U-6Zr-5Nb, and U-6Zr-8Nb alloys in the form of ingots (before hydriding-dehydriding), it can be seen that U-6Zr-2Nb alloy ingots have α U and γ U phases, ingot alloy U-6Zr-5Nb has α U, γ U, and δ 1 (UZr2) phases, U-6Zr-8Nb alloy ingots have α U, γ U, and δ 1 (UZr2) phases. Both U-6Zr-5Nb alloys, both ingots and powders have the same phases of U when in the same conditions. The α U phase appears at an angle of 2θ respectively at 35.4, 38.6, 52.05, and 65.01°, the γ U phase formed at an angle of 2θ respectively at 35.64, 59.8, and the 76° while phase δ 1 (UZr2) is formed at an angle of 2θ at 26.67°. The results are not different from the two forms, namely powder, an ingot, where they produce the α U and γ U phases. Both of these results are also following previous research of making U-Zr-Nb alloys that obtained α U, γ U, and δ 1 (UZr2) phases [15], [16].

IV. CONCLUSION

The results of the analysis of U content in the U-6Zr-2Nb, U-6Zr-5Nb, U-6Zr-8Nb alloy powder samples were 89.307, 85.568, and 83.553 wt%, while the Zr content analysis obtained successive results amounted to 6.220, 5.829, and 6.192 wt%. Meanwhile, in the analysis of Nb in U-6Zr-2Nb alloy powder, U-6Zr-5Nb, U-6Zr-8Nb obtained successive results amounted to 2.023, 5.04, and 8.155 wt%. The recovery results on each U-6Zr-2Nb alloy powder, U-6Zr-5Nb, U-6Zr-8Nb between 96.144 to 99.833%, and a low accuracy rate between 0.002 to 0.029%. From the analysis of impurity elements, it was found that the elements Al, Mn, Cu, and Fe exceeded the requirements. The phase analysis results were obtained for each sample U-6Zr-2Nb, U-6Zr-5Nb, U-6Zr-

8Nb contained α U, γ U, and UO_2 compounds, where the γ U phase was the dominant phase. The highest γ U phase content is found in U-6Zr-5Nb, which is 92.108 wt%, and after the Nb content exceeds 5 wt%, the greater addition of Nb does not increase the number of γ U phases formed.

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